

# Industrial Applications of Molecular Modeling

July 2000

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- Pharmaceuticals & Food Additives
  - Merck – Merck Molecular Force Field for Aqueous Biomolecules
  - Nutrasweet – Rheology of Thickening Agents
  - Unilever – Copolymer Composition of Emulsifiers
- Commodity Chemical
  - Dow – Accurate Thermodynamic Data for Hazard Analysis
  - DuPont – From Process Design to Environmental Fate
- Petrochemical
  - Amoco – Thermodynamics of Intermediates in Design of Catalytic Cycle
- Automotive
  - Lubrizol – Isolating Effect of Individual Components of Complex Product Mixtures, Catalyst Substitution
  - Ford – Improved 3-Way Catalyst Design for Fuel Economy
- Coatings & Colors
  - PPG – Electronic Transitions to Design Photochromics
  - Eastman Kodak – Colors and Electrostatics for Imaging Systems, Acylation Process Improvement

## **Merck – Thomas Halgren**

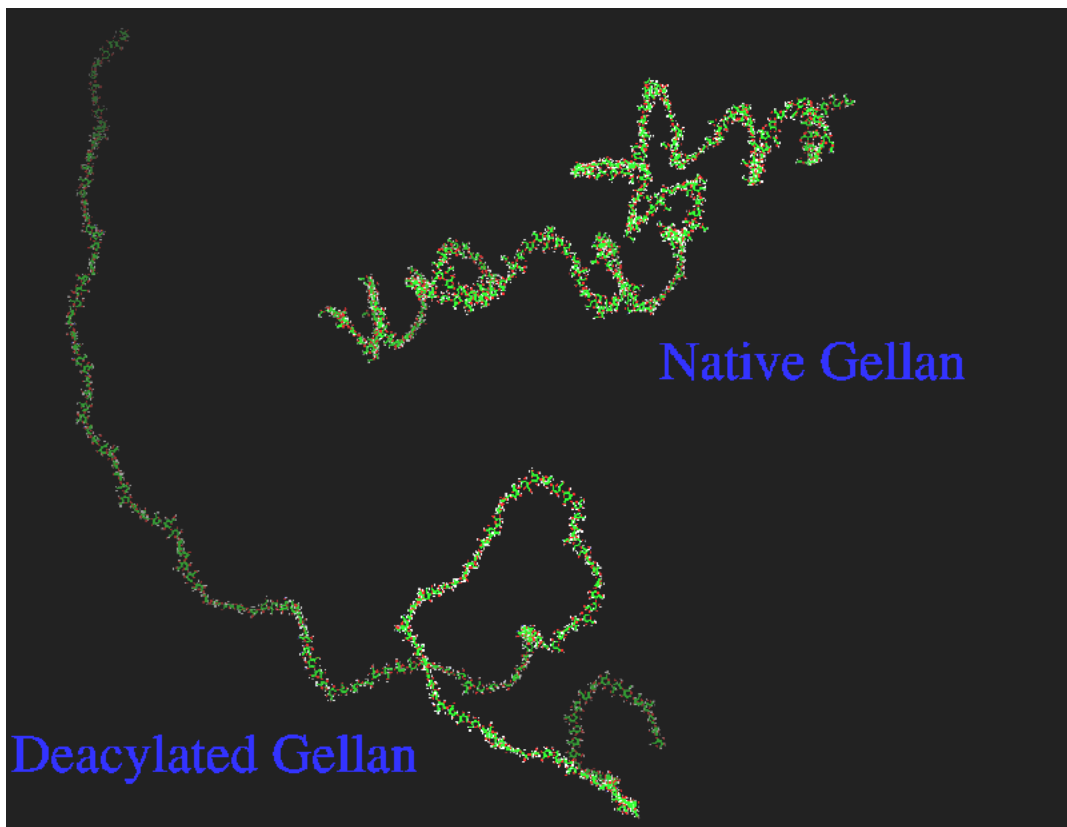
More accurate Molecular Mechanics specific for biomolecules in aqueous environment: drug docking & design, protein folding, CoMFA (Comparative Field Analysis).

- Merck Molecular Force Field (MMFF94)
  - Ab Initio (MP2/6-31G\*) Derived Parameters
  - Class II Force Field
    - Quartic Term (correct for anharmonicity)
    - Cross Term (coupling of vibrations)
- Improved Energies
  - Better Conformational Barriers
  - Better Hydration of Ions (2.9 vs. 8.2 kcal w/ MM2)

## NutraSweet Kelco – Todd Telashek

Control rheology of acylated polysaccharide gelation agents used as thickening agents for food and other applications.

- Model chain extension of chains with various levels of acylation by RIS-MC.

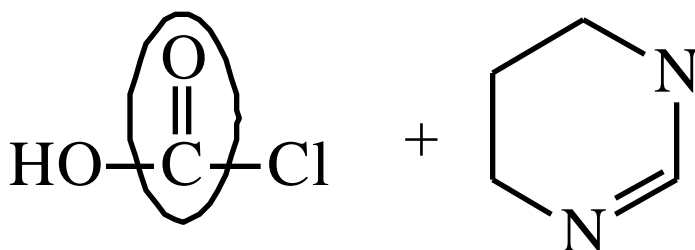


- Trade off brittleness with extended chains at low acylation levels vs elasticity at high acylation levels.

## DOW – Nelson Rondan, David Frurip

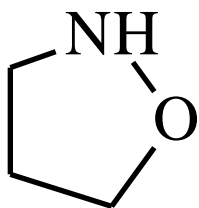
Indirect calculation of thermodynamics for hazard analysis and process design.

- Larger Molecules than DuPont  
Benson Group Contribution Method in CHETAH
- High Level Ab Initio Calculations (G2) to Fill in Missing Groups



25 Hrs. for MP4/6-311G\*\*

- Identified Explosion Hazard by Heat of Decomposition



80% of TNT

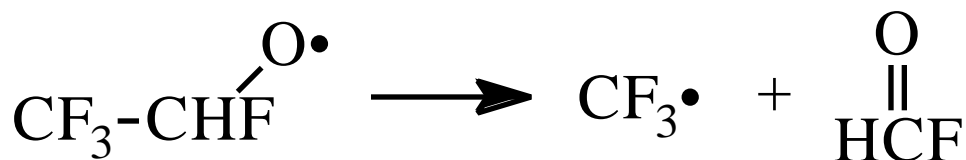
## DuPont – Dave Dixon

Direct calculation of accurate thermodynamics and kinetics for complete product cycle.

- Grand Challenge – design chemical plant from scratch
  - Thermodynamics
  - Kinetics
  - Catalyst Design
  - Process Simulation
  - Fluid Dynamics
  - etc.

- CFC Alternatives – HFC-134A
  - Intermediates in Potential Process for CF<sub>3</sub>CH<sub>2</sub>F
  - NIST Measurement – \$50K + 90 Days
  - MP2/TZP Calculation – \$5k + 7 days

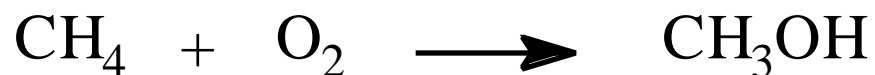
Predict Environmental Fate  
Atmospheric Kinetics



## Amoco – Joe Golab

Find thermodynamic balance in design of catalytic cycle.

- Upgrade Value of Natural Gas: CH<sub>4</sub>



1<sup>st</sup> Step in Oxidative Coupling to Higher Hydrocarbons

- Ta Oxides React with CH<sub>4</sub>

Alkylates Oxide

Doesn't Release Methanol

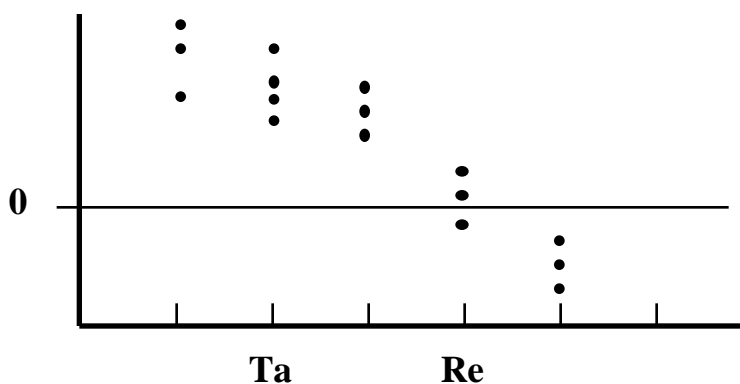
- Model 3rd Row Transition Metal Oxide Families

Oxidation, Coordination, and Spin States

ECP's for Relativistic Atoms

Alkylation Thermodynamics

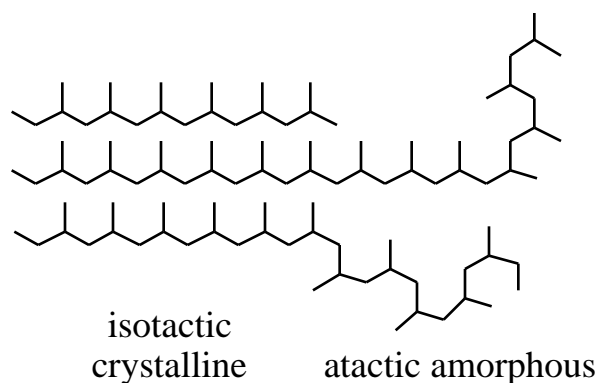
Barrier to Methanol Dissociation



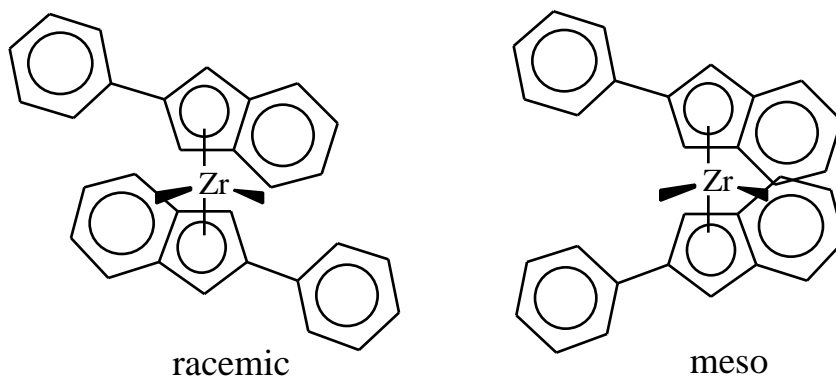
## Petrochemical Companies – Bob Waymouth, Others

Design of new generation olefin polymerization catalysts: elastomeric PP by control of tacticity in blocks.

- Physically Cross-linked Blocks



- Model Multiple Forms of Single Site Catalyst  
Relative Energies by DFT for Transition Metal



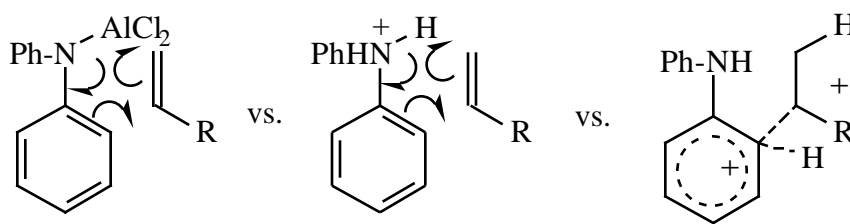
Relative Run Lengths vs. Rate of Change in Catalyst Form  
Stereochemical Control vs. Termination (MW)

## Lubrizol – Anne Chaka

Study intermediates and components of lubricant additives not otherwise isolable: reduce waste, design continuous process, optimize additive.

- Test Cheaper/Safer Replacements for  $\text{AlCl}_3$  &  $\text{BF}_3$  in Additive Synthesis

### Transition State Modeling



Can't combine catalysts in continuous process!

- Optimum Polysulfide for Corrosion Inhibition

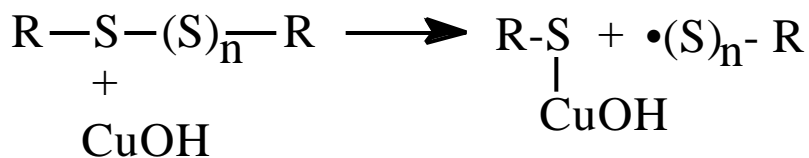
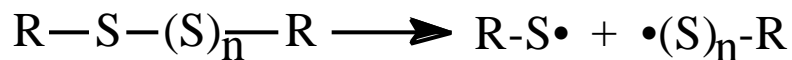


$n = 1-5$  protects Fe in gears

$n > 2$  corrodes Cu alloys

DFT Fast Structure Code

Model Stability w/ & w/o Cu Oxide "Catalysis"

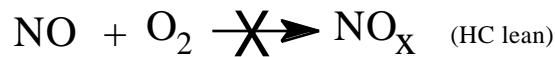
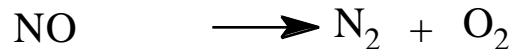
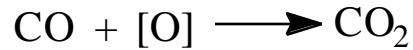




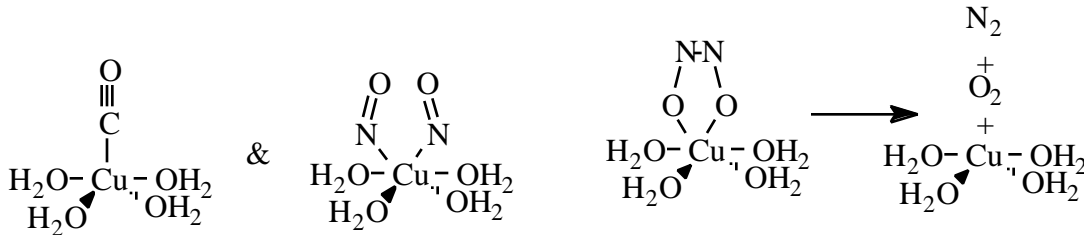
## Ford - Bill Schneider

Design exhaust catalyst for pollution control with lean operation (excess O<sub>2</sub>) for increased fuel economy. Manipulate Cu environment in zeolite to avoid NO<sub>x</sub> in 3-Way Catalyst. Control BaO surface to store NO<sub>x</sub> w/o BaSO<sub>4</sub> in Dual Mode Converter.

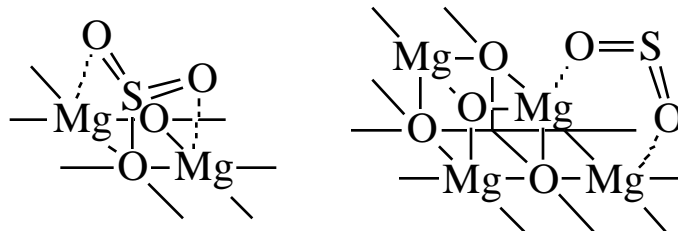
- Catalysts Convert CO and NO<sub>x</sub>



- Model Binding of CO, NO, H<sub>2</sub>O to Supported Cu/ZSM-5 by DFT.



- Model binding of NO<sub>x</sub>, SO<sub>2</sub> & SO<sub>3</sub> on Ba/MgO [100] surface and step with Car-Parinello DFT/MD.



## Procter & Gamble – Bill Laiding

Understand unique property of component in support of deodorant patent.

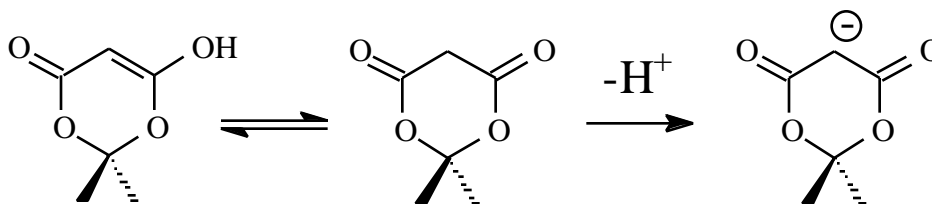
- Meldrum's Acid Unusually Acidic

$\text{pK}_a(\text{DMSO}) = 7.3$

Acyclic (Malonate)  $\text{pK}_a(\text{DMSO}) = 15.9$

- Model Molecules, Tautomers, and Ions to Calculate Acidity

MP2/6-311<sup>++</sup>G<sup>\*\*</sup> for ions



Enol Disfavored

Different Acidities for Axial & Equatorial H

## **PPG – Rick Ross**

Design inorganic pigments & predict photochromic colors and activity.

- **Color an Electronic Process**

  - Promote an Electron into an Unfilled Orbital

  - Energy Difference Determines Wavelength Absorbed

  - ZINDO CI for Good Unfilled Orbital Energies

- **Photochromism**

  - Color vs. MeO- substitution on PNA's

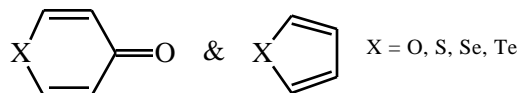
  - Promote Electron from Excited State Instead of Ground State

## Eastman Kodak – John McKelvey, Peter Margl

Checking experimental results, predicting electronics and colors for film and photocopier application, alkylation process improvement.

- Chalcogenic Heterocycles

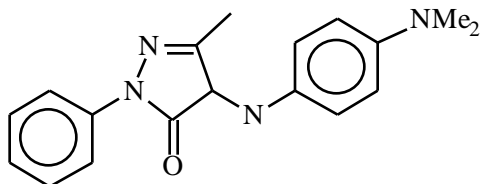
Stabilization of Charge Important for Photocopier  
Anomalous Experimental Se Pyran Dipole Moment



Range of Methods, PM3 Through DFT  
Identified Error in Workup of Lab Data!

- Predicting Color of Pyrazalene Dyes

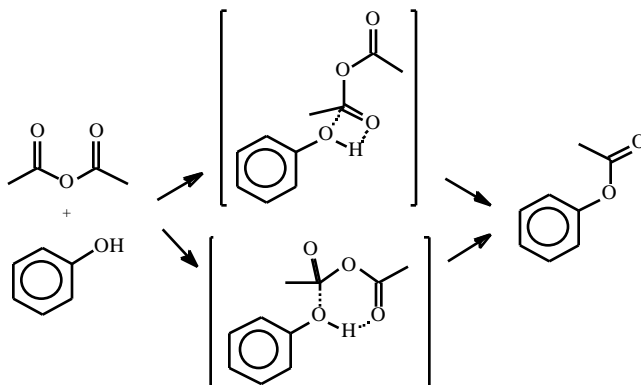
ZINDO for Large Systems



Color & Intensity Controlled by Geometry

- Accelerate Phenol Acylation Process.

Acid vs Base Catalysis of Acyl Transfer Reaction  
B3LYP with SCI-PCM for Solvation.



Acyl Transfer Rate Not Limiting; Solubility Limited!